

Monitoring thermal properties of a composite material used in thermochemical heat storage

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Context and objectives

Conductivity measurement

- in dry conditions
- in wet conditions

Results

- comparison dry/wet conditions
- influence of moisture

Discussion

Conclusions

SoTherCo project

Solar heating for residential buildings:

- Inter-seasonal storage of solar heat using salts and a reversible reaction
- Thermochemical storage

Funding: European Commission, FP7



Low grade heat storage

Thermochemical storage offers high energy density



(S = salt of divalent metal + sulfate, nitrate or halide)

Major structural changes during use phase

S usually entrapped in a porous matrix like silicagel or activated carbon to stabilize the macroscopic structure:

- Reproducibility of storage cycles
- Heat and water transfer

Need of numerical data for heat transport models

- Thermal conductivity $\lambda(T, \%)$ for salt-matrix system with voids between the grains
- Influence of the state of hydration of the salt

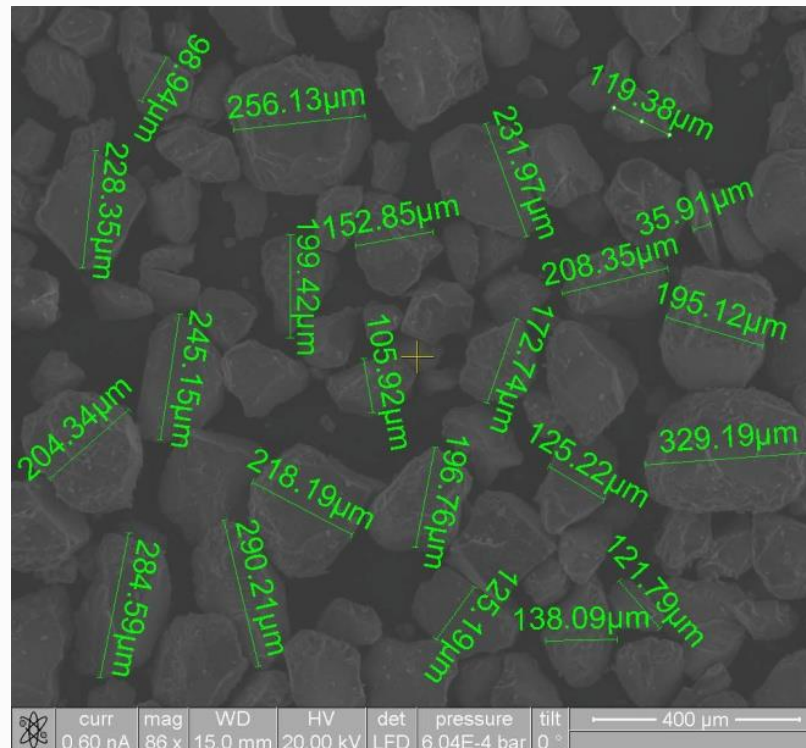
...for a proper automation of these systems

Correlation with mass change?

Is mass transfer more limiting than heat transfer?

Studied material:

- Synthesized at UMONS
- Grains of $\sim 200 \mu\text{m}$ diameter of composite (salt, inorganic matrix)



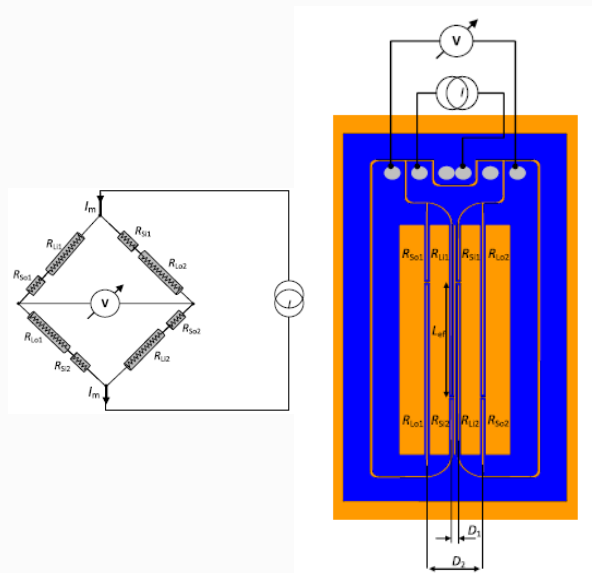
Measurements in dry conditions:

□ Preconditioning of composite (dry state)

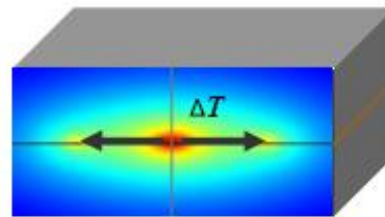
□ Transient Hot Bridge Method (THB)

➤ Theory: from transient Thermal Conductivity Phenomenology

➤ Measurement: Geometry (→ THB sensor), Parameters (→ thermoelectric), THB response

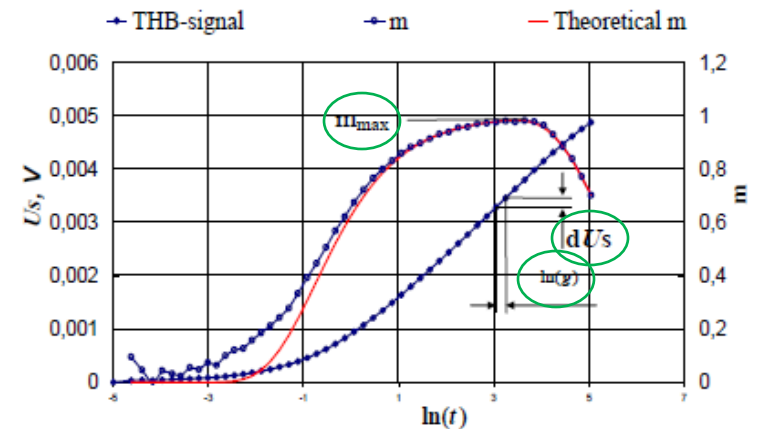


TBH Sensor



Temperature distribution in a sample

$$\lambda = \frac{\alpha \cdot \ln(g)}{4 \cdot \pi \cdot L_{eff}} \cdot \left(\frac{I_m}{2}\right)^2 \cdot m_{max} = f_g \cdot f_{el} \cdot \frac{\ln(g) \cdot m_{max}}{d(U_s)}$$



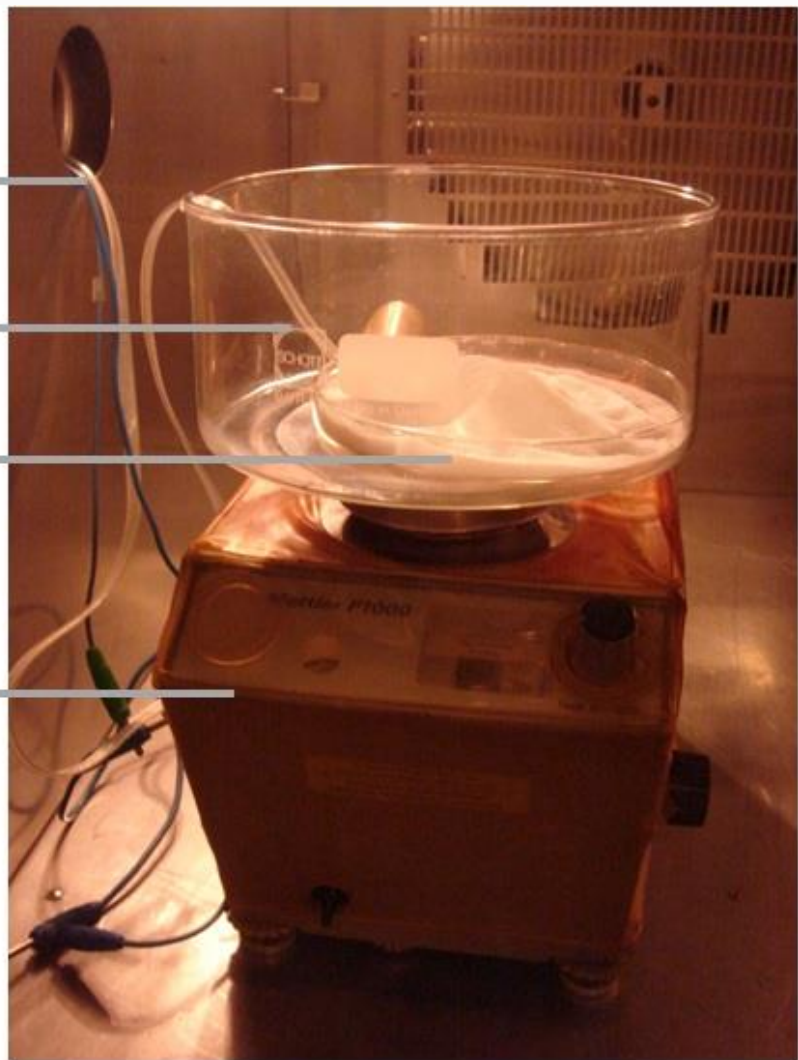
measured THB response, used to examine λ / W/mK, and a / m²/s

Measurements under controlled atmosphere:



VTRK climate chamber
(Heraeus Vötsh)

To computer ←
Vessels to retain the material ←
Sample + sensor ←
Balance ←



In situ measurement of:

- $\lambda(T, \%)$
- Mass

Measurements under controlled atmosphere:

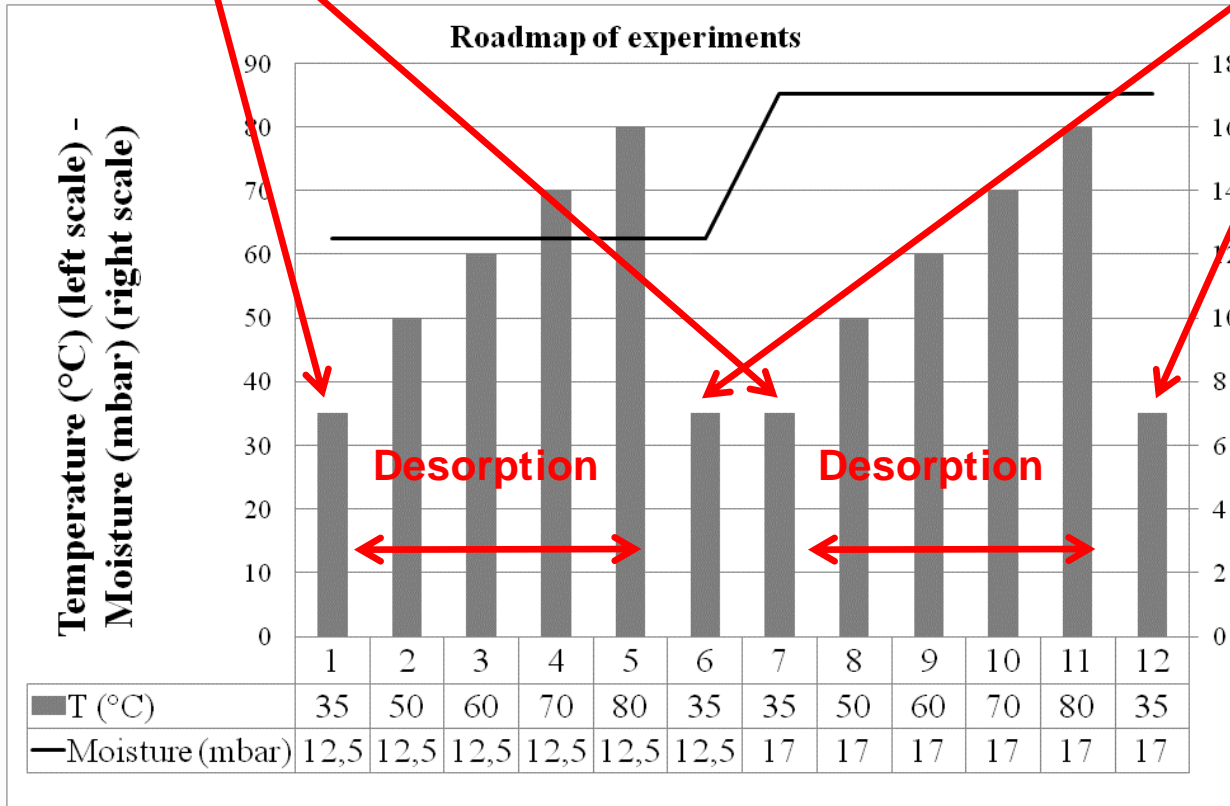
Initial step: preconditioning: hydration

Frequent stirring (slow step)

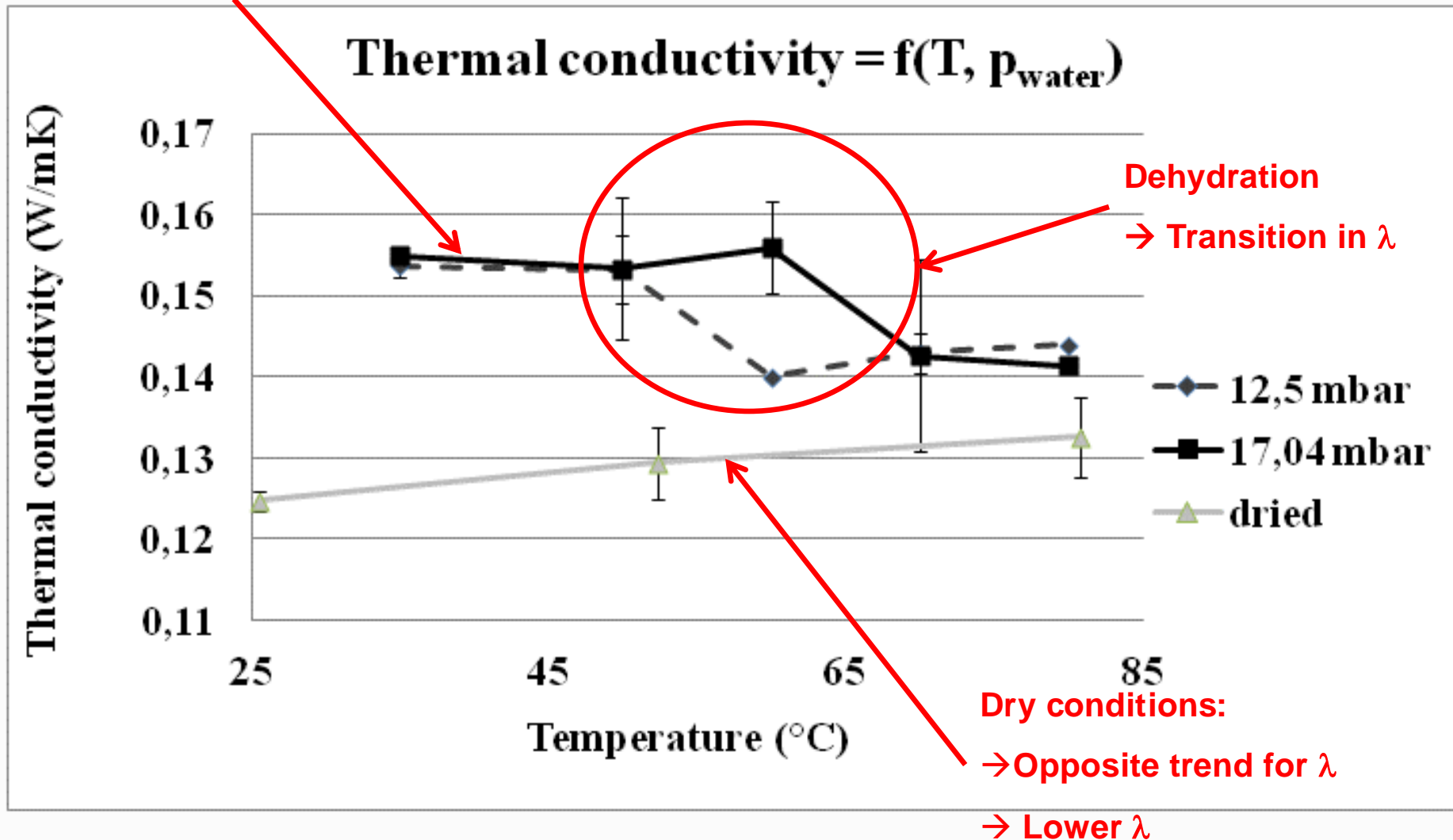
Final step: rehydration

Consistency check

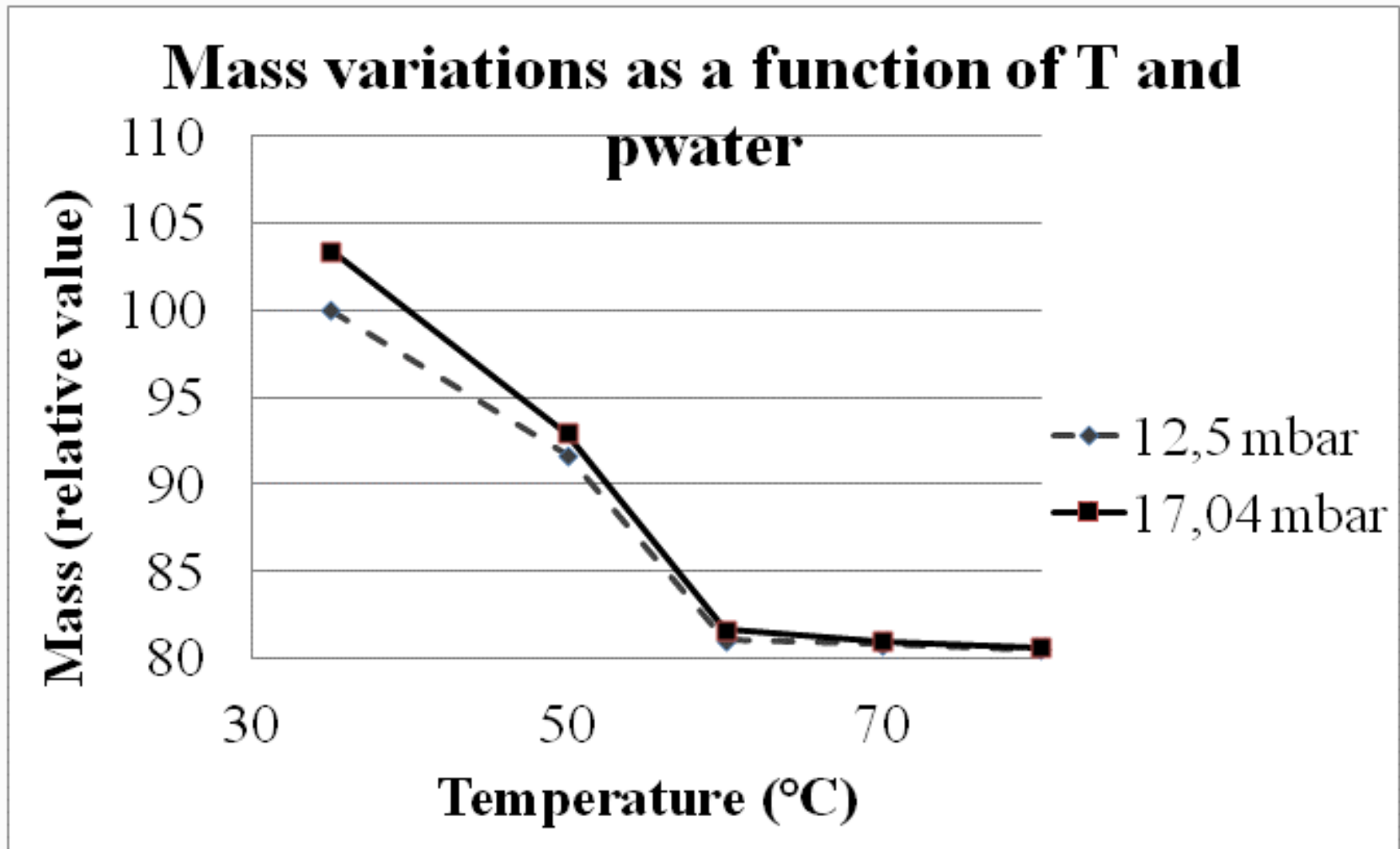
Frequent stirring (slow step)

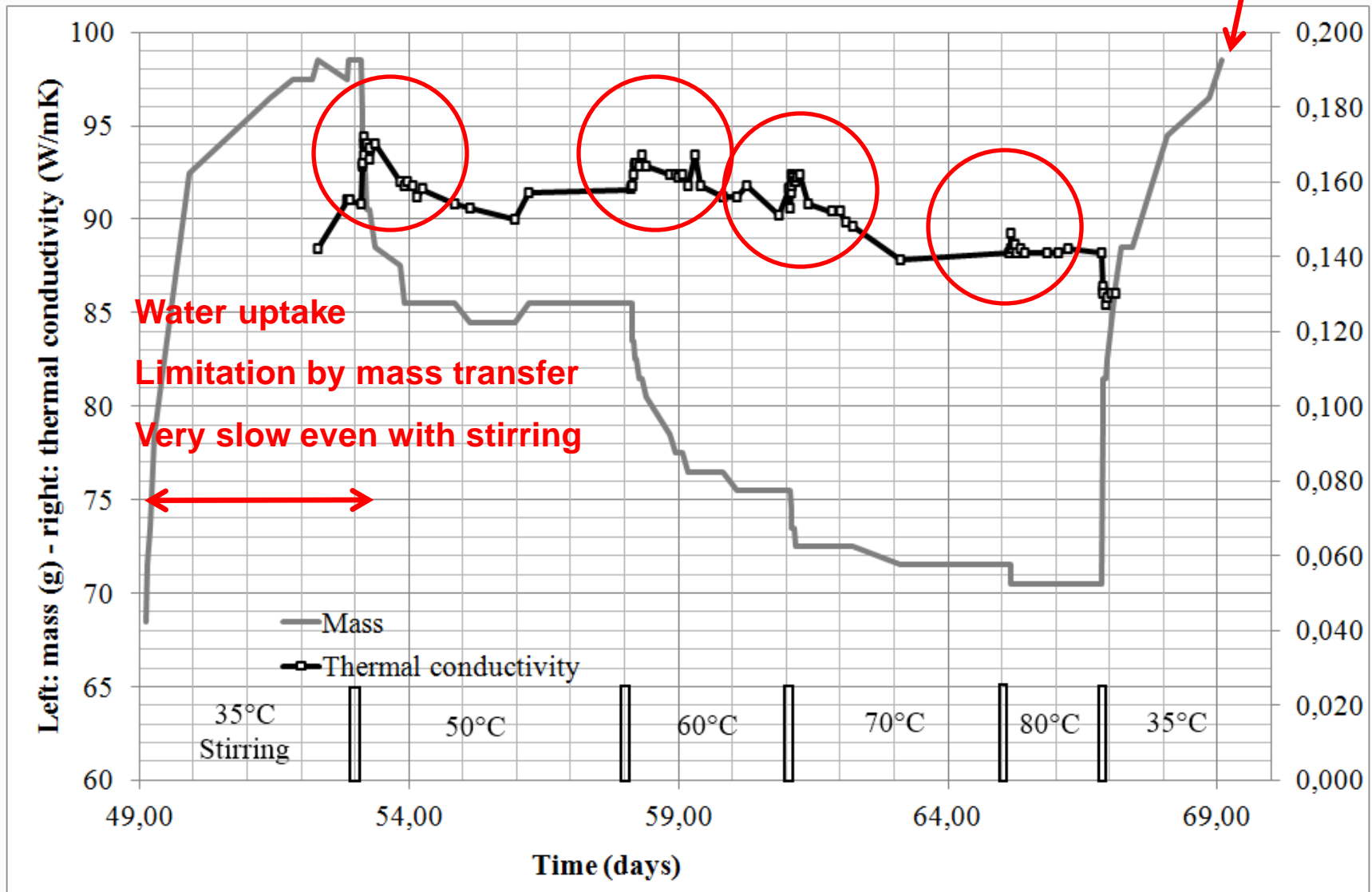


Comparison dry/wet (after transients)

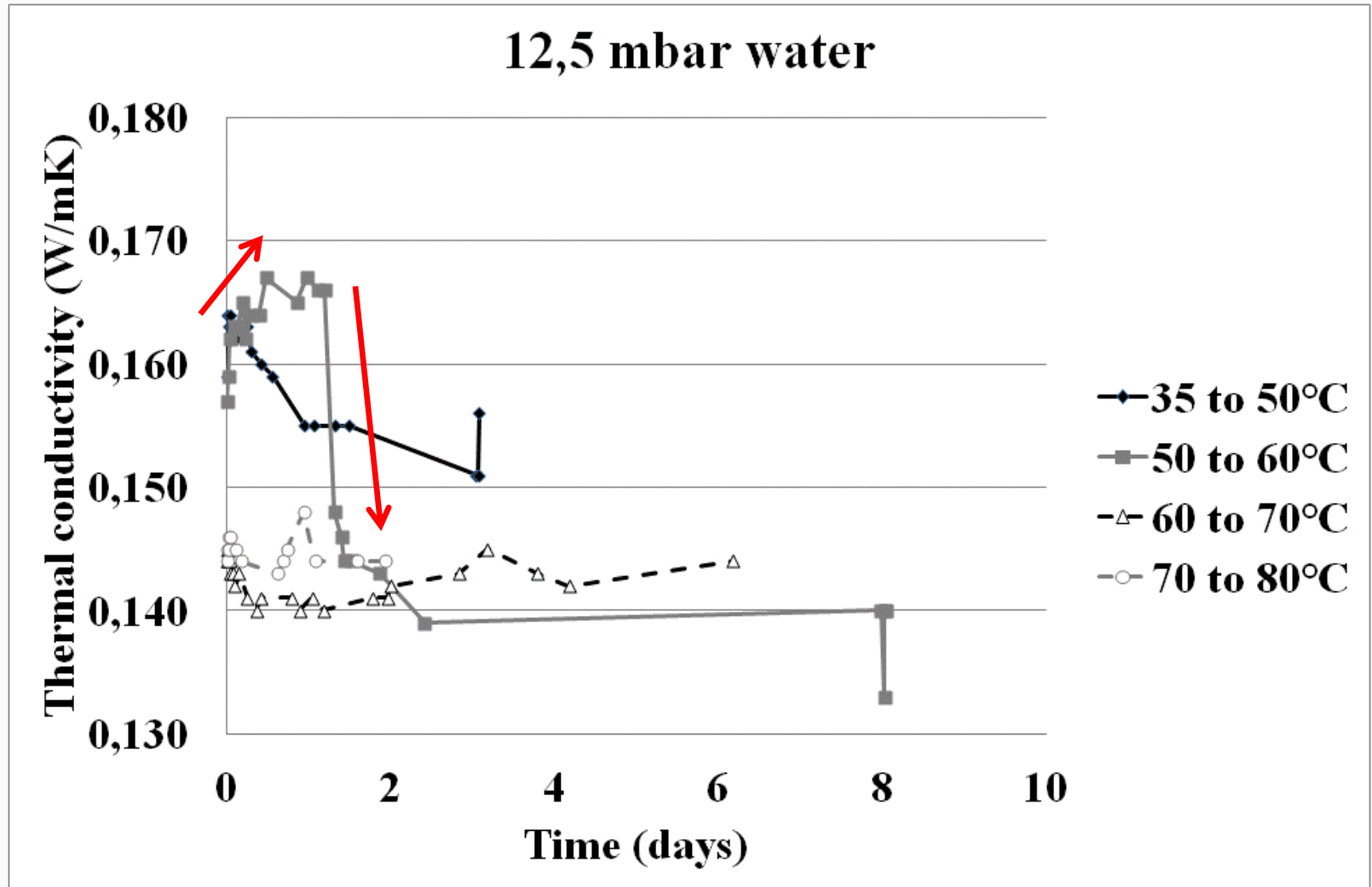
 λ higher in hydrated state

Wet conditions (after transients):

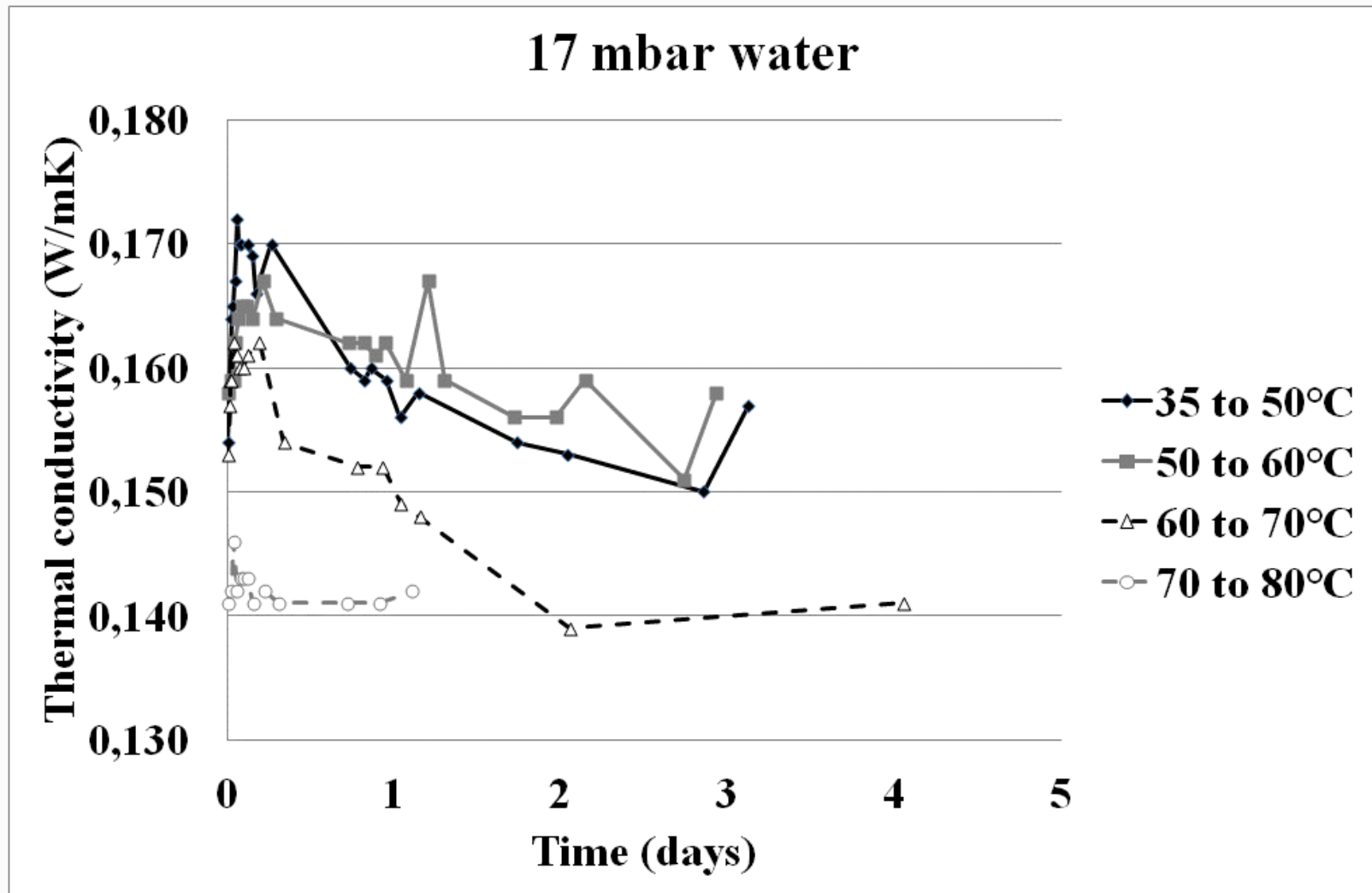


Wet atmosphere: m and λ transients

Wet atmosphere: m and λ transients



Wet atmosphere: m and λ transients



Loss of molecules of H_2O when:

- T increases
- p_{water} decreases

Transition mostly occurs at $35\text{-}60^\circ\text{C}$

Dry material: λ increases with T : typical for ceramics

Wet conditions: λ decreases with T :

- H_2O desorption \rightarrow more voids
- Different hydrates have different intrinsic λ
- Change in convection processes

At 80°C: very little difference between dry and wet experiences: same proportion of H₂O in the salt?

λ during transients: to be interpreted carefully:

- THB method assumes no heat source or sink
- Assumption violated because of water desorption
- % moisture measured by the chamber not representative of % moisture inside the bed

For adapted solutions, see for instance:

Koci J. et al., Proc. Thermophysics, Podkylava, 2013, 55-71

Need of modelling work:

- Existing models focus on the amount of water inside the pores
- Convection around grains neglected
- Conduction at contact points between grains

Experience to assess $\lambda(T, \%)$

Slight increase of $\lambda(T)$ in dry composite

Pronounced decrease of $\lambda(T)$ in wet composite at dehydration

$$\lambda_{\text{wet}}(T) > \lambda_{\text{dry}}(T)$$

Water release generates complex transients

$$0.12 < \lambda < 0.17 \text{ W.m}^{-1}.\text{K}^{-1}$$

Mass transfer at adsorption is a more important limitation for real life application

Perspectives:

- Assessing convection phenomena between the grains
- Assessing conduction in the presence of a endo/exothermic reaction
- Mass transfer limitation to be solved in real-life conditions

Thank you for your attention

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